

A practical enantioselective synthesis of massoialactone via hydrolytic kinetic resolution

Priti Gupta, S. Vasudeva Naidu and Pradeep Kumar*

Division of Organic Chemistry: Technology, National Chemical Laboratory, Pune 411008, India

Received 28 August 2003; revised 30 October 2003; accepted 6 November 2003

Abstract—An efficient enantioselective synthesis of (*R*)- and (*S*)-massoialactone has been achieved. The key steps are the hydrolytic kinetic resolution of a racemic epoxyheptane with (*R,R*)-(salen)-Co^{III}OAc complex and ring-closing metathesis of homoallylic alcohol derived acrylate esters using Grubb's catalyst.

© 2003 Elsevier Ltd. All rights reserved.

δ -Lactones possessing alkyl side chains have attracted much attention from synthetic and medicinal chemists due to their biological activity. One such compound is massoialactone,^{1,2} isolated for the first time from the bark of *Cryptocarya massoia*, by Abe³ in 1937. It has been used for many centuries as a constituent of native medicines. It is a powerful skin irritant and produces systolic standstill in frog heart muscles.¹ This lactone has also been isolated from cane molasses⁴ and jasmine blossoms⁵ as a flavour substance. Later it was isolated from the secretion of the two species of Formicid ants⁶ of the genus *Componotus*, collected in Western Australia.

Various methods for the synthesis of massoialactone (Fig. 1) have been described.^{2,7} The asymmetric syntheses reported in the literature for the natural **1a** and unnatural **1b** isomers of massoialactone either utilize the chiral pool as a starting material⁸ or the chromatographic resolution of the diastereomeric derivative of the lactone precursor.⁹ A recent report describes the synthesis via asymmetric allylboration of an aldehyde with β -allyldiisopinocampheylborane.¹⁰ As part of our research program aimed at developing enantioselective syntheses of naturally occurring lactones¹¹ and amino alcohols¹² we recently reported the synthesis of (*S*)-massoialactone using the Sharpless asymmetric dihydroxylation approach.^{11a} However the enantiomeric purity of the diol obtained was not high due to the terminal olefin employed as a substrate in the dihydroxylation

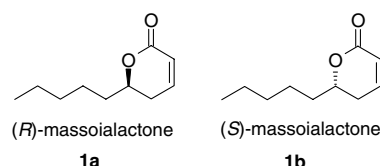
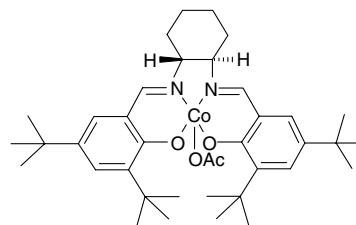


Figure 1.

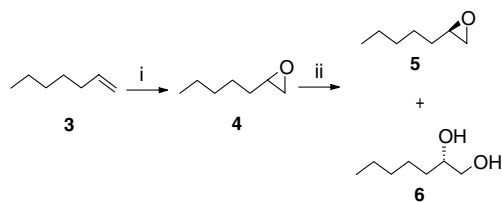
step. Herein we report a new and highly enantioselective synthesis of (*R*)- and (*S*)-massoialactone using Jacobsen's hydrolytic kinetic resolution (HKR) of a terminal epoxide.¹³ The HKR method uses readily accessible cobalt-based chiral salen complex **2** (Fig. 2) as catalyst and water as the only reagent to resolve a racemic epoxide into the enantiomerically enriched epoxide and diol in high enantiomeric excess. These advantages have made it a very attractive asymmetric synthetic tool.



(*R,R*)-SalenCo^{III}OAc complex **2**

Figure 2.

* Corresponding author. Tel.: +91-20-589-3300x2050; fax: +91-20-589-3614; e-mail: tripathi@dalton.ncl.res.in

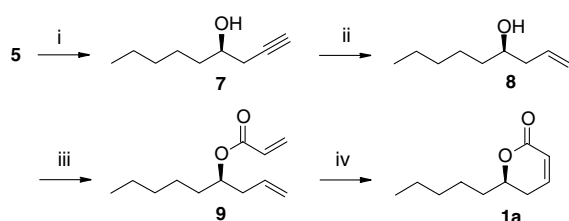


Scheme 1. Reagents and conditions: (i) *m*-CPBA, CH₂Cl₂, 0 °C to rt, 10 h, 92%; (ii) *R,R*-salen-Co(OAc) (0.5 mol%), distd H₂O (0.55 equiv), 0 °C, 16 h, (45% for **5**, 43% for **6**).

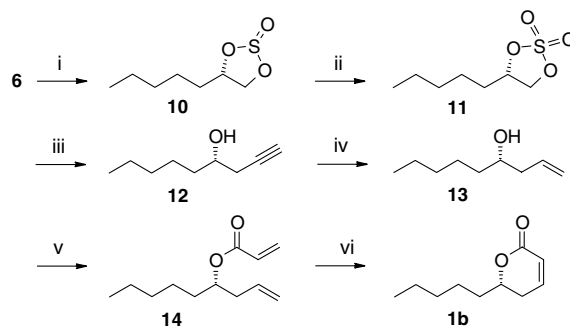
The racemic epoxide **4**, a substrate for HKR was prepared from commercially available 1-heptene **3** using *m*-CPBA. The HKR was performed on **4** with (*R,R*)-salen-Co(OAc) complex **2** (0.5 mol%) and water (0.55 equiv) to give the *R*-epoxide **5** in 45% yield with >99% ee,¹⁴ [α]_D²⁵ +9.6 (*c* 1, CHCl₃) [lit.¹⁵ +9.8 (*c* 1, CHCl₃)] and the *S*-diol **6** in 43% yield with 99.5% ee,¹⁶ [α]_D²⁵ -15.9 (*c* 1.67, EtOH) [lit.^{8a} [α]_D²² -15.2 (*c* 1.67, EtOH)] (Scheme 1).

The synthesis of (*R*)-massoialactone **1a** started from the enantiomerically enriched epoxide **5** as illustrated in Scheme 2. Thus opening of **5** with an excess of lithium acetylide followed by partial hydrogenation of the resultant acetylene **7** with Lindlar's catalyst furnished the homoallylic alcohol **8**. Compound **8** was esterified with acryloyl chloride in the presence of triethylamine to afford **9** in 89% yield. The subsequent ring-closing metathesis¹⁷ in dichloromethane under reflux in high dilution conditions using the first generation Grubbs's catalyst, bis(tricyclohexylphosphine)benzylidene ruthenium(IV) dichloride and catalytic amount of Ti(*i*-PrO)₄ afforded (*R*)-massoialactone in 84% yield, [α]_D²⁵ -115.2 (*c* 1, CHCl₃) [lit.¹⁰ [α]_D²⁹ -113.6 (*c* 1.36, CHCl₃)]. The physical and spectroscopic data were in full agreement with the literature.^{8c}

Scheme 3 summarizes the synthesis of (*S*)-massoialactone **1b** from the diol **6**. Thus treatment of **6** with thionyl chloride in the presence of triethylamine gave the cyclic sulfite **10**, which was further oxidized using NaIO₄ and a catalytic amount of ruthenium trichloride to furnish the corresponding cyclic sulfate **11** in essentially quantitative yield.¹⁸ The essential feature of our synthetic strategy shown in Scheme 3 was based on the presumption



Scheme 2. Reagents and conditions: (i) LiC≡CH-ethylene diamine, DMSO, rt, 12 h, 86%; (ii) H₂, Pd/BaSO₄, quinoline, benzene, 1 bar, rt, 0.5 h, 92%; (iii) acryloyl chloride, Et₃N, CH₂Cl₂, 0 °C, 5–6 h, 89%; (iv) (PCy₃)₂ Ru(Cl)₂=CH-Ph (20 mol%), CH₂Cl₂, Ti(*i*-PrO)₄, reflux, 12 h, 84%.



Scheme 3. Reagents and conditions: (i) SOCl₂, Et₃N, CH₂Cl₂, 0 °C, 20 min, 99%; (ii) RuCl₃, NaIO₄, CCl₄-MeCN-H₂O; 2:2:3, 0 °C, 2 h, 100%; (iii) LiC≡CH-ethylene diamine, DMSO, 0 °C to rt, 10 h, 80%; (iv) H₂, Pd/BaSO₄, quinoline, benzene, 1 bar, rt, 0.5 h, 86%; (v) acryloyl chloride, Et₃N, CH₂Cl₂, 0 °C, 5–6 h, 84%; (vi) (PCy₃)₂Ru(Cl)₂=CH-Ph (20 mol%), CH₂Cl₂, Ti(*i*-PrO)₄, reflux, 12 h, 85%.

that the nucleophilic opening of the cyclic sulfate **11** would occur in a regioselective manner at the terminal carbon. Indeed the cyclic sulfate **11** on treatment with lithium acetylide furnished the desired alcohol **12**, which on hydrogenation followed by ring-closing metathesis afforded the target molecule **1b**, [α]_D²⁵ +110.1 (*c* 2.0, CHCl₃) [lit.⁹ [α]_D^{22.6} +109.6 (*c* 2, CHCl₃)]. The physical and spectroscopic data were in full agreement with the literature.^{8c}

In conclusion we have demonstrated that the enantioselective synthesis of both the isomers of massoialactone can be accomplished using hydrolytic kinetic resolution of a racemic epoxide and ring-closing metathesis. The synthetic strategy described has significant potential for further extension to a variety of other 6-substituted chiral lactones, which serve as important synthons for several naturally occurring and biologically active molecules. Currently studies are in progress in this direction.

Acknowledgements

P.G. and S.V.N. thank UGC and CSIR New Delhi for financial assistance, respectively. We are grateful to Dr. M. K. Gurjar for his support and encouragement. This is NCL Communication No. 6652.

References and notes

- Meijer, Th. M. *Rec. Trav. Chim. Pays-Bas*. **1940**, *59*, 191–201.
- Crombie, L. *J. Chem. Soc.* **1955**, 1007–1025, 2535.
- Abe, S. *J. Chem. Soc. Jpn.* **1937**, *58*, 246–251.
- Hashijune, T.; Kikuchi, N.; Sasaki, Y.; Sakata, I. *Agric. Biol. Chem.* **1998**, *32*, 1306–1309.
- Kaiser, P.; Lamparsky, D. *Tetrahedron Lett.* **1976**, 1659–1660.
- Cavill, G. W. K.; Clark, D. V.; Whitfield, F. B. *Aust. J. Chem.* **1968**, *21*, 2819–2823.

7. (a) Nobuhara, A. *J. Agric. Biol. Chem.* **1968**, *32*, 1016–1020; (b) Abe, S.; Sato, K. *Bull. Chem. Soc. Jpn.* **1956**, *29*, 88–90.
8. (a) Mori, K. *Agric. Biol. Chem.* **1976**, *40*, 1617–1619; (b) Asaoka, M.; Hayashibee, S.; Sonoda, S.; Takei, H. *Tetrahedron Lett.* **1990**, *31*, 4760–4764; (c) Bannet, F.; Knight, D. W. *Heterocycles* **1989**, *29*, 639–642; (d) Minami, T.; Moriyama, A.; Hanaoka, M. *Synlett* **1995**, 663–665.
9. Pirkle, W. H.; Adams, P. E. *J. Org. Chem.* **1980**, *45*, 4117–4121.
10. Ramachandran, P. V.; Reddy, M. V. R.; Brown, H. C. *Tetrahedron Lett.* **2000**, *41*, 583–586.
11. (a) Pais, G. C. G.; Fernandes, R. A.; Kumar, P. *Tetrahedron* **1999**, *55*, 13445–13450; (b) Fernandes, R. A.; Kumar, P. *Tetrahedron: Asymmetry* **1999**, *10*, 4349–4356; (c) Fernandes, R. A.; Kumar, P. *Eur. J. Org. Chem.* **2002**, 2921–2923; (d) Kandula, S. V.; Kumar, P. *Tetrahedron Lett.* **2003**, *44*, 6149–6151.
12. (a) Fernandes, R. A.; Kumar, P. *Eur. J. Org. Chem.* **2000**, 3447–3449; (b) Pandey, R. K.; Fernandes, R. A.; Kumar, P. *Tetrahedron Lett.* **2002**, *43*, 4425–4426; (c) Fernandes, R. A.; Kumar, P. *Tetrahedron Lett.* **2000**, *41*, 10309–10312; (d) Naidu, S. V.; Kumar, P. *Tetrahedron Lett.* **2003**, *44*, 1035–1037; (e) Kandula, S. V.; Kumar, P. *Tetrahedron Lett.* **2003**, *44*, 1957–1958; (f) Gupta, P.; Fernandes, R. A.; Kumar, P. *Tetrahedron Lett.* **2003**, *44*, 4231–4232.
13. (a) Tokunaga, M.; Larrow, J. F.; Kakiuchi, F.; Jacobsen, E. N. *Science* **1997**, *277*, 936–938; (b) Schaus, S. E.; Branalt, J.; Jacobson, E. N. *J. Org. Chem.* **1998**, *63*, 4876–4877.
14. The enantiomeric purity of the epoxide **5** was estimated to be >99% by chiral GC analysis of the 1-azido-2-trimethylsilyloxyheptane derivative using cyclodex-B (30 m × 0.25 mm i.d.) at 50 °C for 1 min, 20 °C/min to 220 °C.
15. Haase, B.; Schneider, M. P. *Tetrahedron: Asymmetry* **1993**, *4*, 1017–1026.
16. For the measurement of enantiomeric excess, the diol **6** was converted into its dibenzoate. The enantiomeric purity of the dibenzoate was estimated to be 99.5% by chiral HPLC analysis (Chiralcel OD, 98:2 petroleum ether-*i*PrOH, 1 mL/min, 240 nm).
17. For reviews on ring-closing metathesis, see: (a) Grubbs, R. H.; Chang, S. *Tetrahedron* **1998**, *54*, 4413–4450; (b) Prunet, J. *Angew. Chem., Int. Ed.* **2003**, *42*, 2826–2830.
18. For reviews on cyclic sulfites/cyclic sulfates, see: (a) Lohray, B. B. *Synthesis* **1992**, 1035–1052; (b) Byun, H.-S.; He, L.; Bittman, R. *Tetrahedron* **2000**, *56*, 7051–7091.